

and locating the sapogenins with 50%  $\text{H}_2\text{SO}_4$ .<sup>7</sup> The keto-sapogenins were isolated by PLC on air dried silica gel G layers, 500  $\mu\text{m}$ , with double development in cyclohexane-EtOAc- $\text{H}_2\text{O}$  (600:400:1). A sample of the keto-sapogenin mixture was reduced by the Huang-Minlon modification of the Wolff-Kishner method.<sup>13</sup> Acetates of the compounds formed were prepared by refluxing the sapogenins with  $\text{Ac}_2\text{O}$  for 30 min and sapogenin trifluoroacetates were prepared by the method of Bennett and Heftmann.<sup>7</sup> The sapogenin acetates and trifluoroacetates were examined on activated silica gel G layers, 250  $\mu\text{m}$ , using  $\text{CHCl}_3$ -toluene (9:1). The three keto-sapogenins were separated from each other by PLC on silica gel G layers, 500  $\mu\text{m}$ , developing  $4 \times$  in *n*-hexane-EtOAc (6:1).

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<sup>13</sup> HUANG-MINLON (1946) *J. Am. Chem. Soc.* **68**, 2487.

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## THE ESSENTIAL OIL IN *SCIRPUS AMERICANUS*\*

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**Key Word Index**—*Scirpus americanus*; Cyperaceae; essential oil; phenol; 2 phenyl ethanol; p-tolualdehyde; cuminyl alcohol.

The Mississippi salt marsh is an irregularly flooded estuary dominated by the needle-rush. *Juncus roemerianus* (Juncaceae), the giant cordgrass *Spartina cynosuroides* (Graminae), and *Scirpus americanus* (Cyperaceae). Preliminary studies by Odum<sup>2</sup> include those on primary production and decomposition of *Scirpus americanus*, and the food value of this species to marsh and estuarine organisms. To our knowledge, there is no report of a detailed study on the organic constituents of *S. americanus*. This communication on the essential oil of *S. americanus* is a part<sup>3,4</sup> of a continuing chemoecological study of the Mississippi salt marsh.

An investigation of the essential oil of the marsh grass, *S. americanus*, by combined GC-MS resulted in the identification of 40 compounds that comprise 78.8% of the total oil.

\* Part IV in the series "Constituents of Marsh Grass". For Part III, see Ref. 1.

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<sup>1</sup> BHATTACHARYYA, J., MODY, N. V., HEDIN, P. A. and MILES, D. H. (1974) *Tetrahedron Letters*.

<sup>2</sup> ODUM, E. P. and DE LA CRUZ, A. A. (1963) *AIBS Bull.* **13**, 39.

<sup>3</sup> MILES, D. H., MODY, N. V., MINYARD, J. P. and HEDIN, P. A. (1973) *Phytochemistry* **12**, 1399.

<sup>4</sup> MODY, N. V., BHATTACHARYYA, J., MILES, D. H. and HEDIN, P. A. (1974) *Phytochemistry*. In press.

The structures of 15 aliphatic and aromatic hydrocarbons, 9 aldehydes, 2 ketones, 12 alcohols, phenols and ethers, and benzothiazole are proposed. Phenol, 2-phenyl ethanol, *p*-tolualdehyde, cuminyl alcohol, (4-methyl-pent-3-enyl) cyclohexene-1-carboxaldehyde, 2,4-ditert-butylphenol and *n*-heptadecane are the major components of the oil. Benzothiazole was the only heterocyclic compound found.

### EXPERIMENTAL

*Isolation of the essential oil.* Fresh marsh grass (1 kg) was harvested from Bay St. Louis, Mississippi, and stored below 0° until it was chopped and steam distilled in an all glass system for about 3 hr. The distillate was extracted with Et<sub>2</sub>O, dried (Na<sub>2</sub>SO<sub>4</sub>) and the solvent removed *in vacuo* at 35° to give 0.2 g of oil. Yield = 0.02% [calculated on the basis of fr. plant].

*Column chromatography.* The oil was chromatographed on a 2 × 25 cm jacketed Florisil column cooled with ice water to prevent column cracking. The column was eluted with 200 ml portions of pentane; 10% Et<sub>2</sub>O in pentane and 100% Et<sub>2</sub>O. The separations were monitored by TLC, and the components were located by heating the developed plate after spraying with 3% vanillin in 0.5% conc H<sub>2</sub>SO<sub>4</sub> in MeOH.

*Analytical GC-MS.* The three fractions were introduced separately into a Hewlett-Packard 5930 quadrupole mass spectrometer interfaced with a 5700 Å gas chromatograph from a 74 m × 0.75 mm capillary gas chromatographic column coated with OV-17. Carrier gas flow was 8.0 ml/min N<sub>2</sub>. The GLC unit was programmed from 120° to 160° at 1°/min. The final temperature was maintained for 20 min. MS were obtained at 70 ev. Fragment

TABLE 1. ANALYSIS OF THE ESSENTIAL OIL IN *Scripus americanus*

Compound	I <sub>K</sub> OV-17*	Fragmentation†	(%)‡
<b>Hydrocarbons</b>			
Toluene	978	R	Trace
<i>m</i> -Xylene	1040	R	0.4
<i>p</i> -Xylene	1055	R	0.2
<i>n</i> -Decane	1010	R	0.2
<i>n</i> -Propylbenzene	1090	R	3.2
Isopropylbenzene	1120	R	2.5
Trimethylbenzene	1130	R	3.2
3-Ethyl- <i>o</i> -xylene	1210	R	0.6
Indane	1230	R	0.1
Tetramethylbenzene	1240	R	Trace
<i>n</i> -Tridecane	1300	R	1.0
<i>n</i> -Tetradecane	1400	R	0.8
<i>n</i> -Pentadecane	1505	R	3.0
<i>n</i> -Hexadecane	1600	R	0.6
<i>n</i> -Heptadecane	1700	R	4.2
<b>Carbonyl compounds</b>			
C <sub>6</sub> H <sub>8</sub> O aldehyde	1030	43, 44, 41, 57, 55, 96	Trace
Benzaldehyde	1160	R	0.2
C <sub>7</sub> H <sub>14</sub> O aldehyde	—	43, 41, 57, 55, 70, 114	1.0
2-Furaldehyde	1062	R	0.2
4-Methylcyclohexanone	1190	R	0.2
C <sub>8</sub> H <sub>12</sub> O ketone	—	68, 41, 39, 43, 55, 124	Trace
Phenyl acetaldehyde	1250	R	1.8
<i>p</i> -Tolualdehyde	1515	R	16.1
Geranial	—	R	0.5
(4-Methyl-pent-3-enyl) cyclohexadiene-1- carboxaldehyde <sup>3,4</sup>	1540	R	2.5
(4-Methyl-pent-3-enyl) cyclohexene-1-carboxaldehyde <sup>3</sup>	1570	R	7.9
<b>Alcohols, phenols and miscellaneous</b>			
3-Hexene-1-ol	—	R	0.1
Phenol	1230	R	6.2

TABLE 1. *contd.*

Compound	I <sub>k</sub> OV-17*	Fragmentation†	(%)‡
Benzyl alcohol	1262	R	0.2
<i>o</i> -Methoxyphenol	1312	R	2.7
2-Phenylethanol	1450	R	4.0
Cumyl alcohol	1560	R	8.0
Butyl phenol	—	119, 150, 135, 91, 107, 150	0.1
C <sub>10</sub> H <sub>16</sub> O phenol	—	43, 41, 69, 117, 107, 152	1.3
<i>p</i> -Tert-butylphenetole	1640	R	1.0
2,4-Ditert-butylphenol	1705	R	3.1
C <sub>10</sub> H <sub>14</sub> O	—	43, 41, 53, 81, 69, 150	Trace
C <sub>10</sub> H <sub>16</sub> O	—	41, 43, 67, 55, 81, 152	0.6
Benzothiazole	1475	R	1.1
C <sub>13</sub> H <sub>20</sub> O	—	55, 57, 163, 177, 79, 192	Trace

\* KOVATES, E. (1961) *Anal. Chem.* **181**, 351.

† The five most intense fragment ion values (*m/e*) arranged in order of decreasing relative abundance with the proposed parent ion presented sixth: R means that the spectrum corresponded with literature values.<sup>3-6</sup>

‡ Percentage of total oil.

ion values were compared with literature values.<sup>3-6</sup> GLC retention times are presented as Kovats indices<sup>7</sup> (I<sub>k</sub>). Estimates of the per cent content of each component in the oil were made by peak triangulation of the maxima of the GLC profile trace.

<sup>5</sup> STENHAGEN, E., ABRAHAMSSON and MCLAFFERTY, F. W. eds. (1969) *Atlas of Mass Spectral Data*. Interscience, New York.

<sup>6</sup> CORNU, A. and MASSOT, R. (1966) *Compilation of Mass Spectral Data*. Heyden. London.

<sup>7</sup> KOVATS, E. (1961) *Anal. Chem.* **181**, 351.

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## CORN BUD ESSENTIAL OIL

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**Key Word Index**—*Zea mays* L.; gramineae; corn; bud essential oil.

As a part of current studies of the essential oils of plants of commercial and economic importance,<sup>1,2</sup> the steam distillate of the whorl (bud) of corn, *Zea mays* L., in the mid-whorl stage was subjected to GC-MS analysis.

<sup>1</sup> MINYARD, J. P., TUMLINSON, J. H., THOMPSON, A. C. and HEDIN, P. A. (1966) *J. Agr. Food Chem.* **14**, 332.

<sup>2</sup> HANNY, B. W., THOMPSON, A. C., GUELDNER, R. C. and HEDIN, P. A. (1973) *J. Agr. Food Chem.* **21**, 1004.